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NC	OIDA	INSTITUTE OF ENGINEERING A	ND TECHNO	OLOG	Y, G1	REAT	ER N		DA
110		(An Autonomous Institute Aff			,				
		B.Te	_						
		SEM: III - THEORY EXAMI				024)			
Tin	3 I	Subject: Materials Scie Hours	ence and Engi	neerir	ıg	Mo	v Ma	arlz	s: 100
		structions:				IVIa	IX. IVI	11 K	5. 100
		y that you have received the question p	aper with the	correc	t cour	se, coa	le, bra	ıncl	h etc.
		stion paper comprises of three Section	-						
		MCQ's) & Subjective type questions.							
		n marks for each question are indicated	_		of ea	ch que	stion.		
		your answers with neat sketches where suitable data if necessary.	ever necessar <sub>.</sub>	y <b>.</b>					
		ly, write the answers in sequential orde	er.						
	v	should be left blank. Any written mater		ınk she	et will	not be	?		
evalu	ated/c	hecked.							
SEC <sub>1</sub>	rion.	<u>-A</u>							20
1. Att	empt	all parts:-			<b>\</b>				
1-a.	T	he elastic stress strain behavior of rubb	per is (CO1)						1
	(a)	Liner		)					
	(b)	Nonliner	1						
	(c)	Plastic							
	(d)	No fixed relationship							
1-b.	C	crystal structure of austenite is (CO1)							1
	(a)	Body centered cubic							
	(b)	Face centered cubic							
	(c)	Hexagonal closed packed							
	(d)	Body centered tetragonal							
1-c.	G	Sibb's phase rule is given by $(F = no. of$	CDOF, C = no	of co	mpone	ents, P	= no.	of	1
	p]	hases) (CO2)							
	(a)	F = C + P							
	(b)	F = C + P + 2							
	(c)	F = C - P - 2							
	(d)	F = C - P + 2							
1-d. Pearlite consists of (CO2)								1	
	(a)	6.67% C and 93.33% ferrite							
	(b)	12% Fe and 87% cementite							

	(c)	13% C and 87% ferrite		
	(d)	13% cementite and 87% ferrite		
1-e.	T	he lattice diffusion is caused by (CO3)	1	
	(a)	Grain boundaries		
	(b)	Screw dislocations		
	(c)	Point imperfections		
	(d)	Twins		
1-f.	Which of the following is not a stage of annealing? (CO3)			
	(a)	Heating		
	(b)	Soaking		
	(c)	Tempering		
	(d)	Quenching		
1-g.	C	arbon nano tubes are also called as (CO4)	1	
	(a)	Bucky tubes		
	(b)	Bulky tubes		
	(c)	Bulk tubes		
	(d)	Buck balls		
1-h.	St	crong and ductile materials (CO4)	1	
	(a)	Polymers		
	(b)	Ceramics		
	(c)	Metals		
	(d)	Semiconductors		
1-i.		Thich among the following helps us in getting a three-dimensional picture of the becimen? (CO5)	1	
	(a)	Transmission Electron Microscope		
	(b)	Scanning Electron Microscope		
	(c)	Compound Microscope		
	(d)	Simple Microscope		
1-j.	X	-Ray can be deflected by (CO5)	1	
	(a)	Electric field		
	(b)	Magnetic field		
	(c)	Electromagnetic field		
	(d)	None of the fields		
2. Att	empt a	all parts:-		
2.a.	D	raw stress-strain diagram for cast iron. (CO1)	2	
2.b.		That are the four solid phases present in the iron-iron carbide phase diagram?	2	
2.c.	W	That are the objective of normalizing? (CO3)	2	

2.d.	What are the advantages of composite materials. (CO4)	2
2.e.	How SEM is differ from FESEM? (CO5)	2
<b>SECTIO</b>	$\mathbf{N}\mathbf{B}$	30
3. Answe	er any <u>five</u> of the following:-	
3-a.	What are the factors affecting fatigue properties of materials, explain in brief. (CO1)	6
3-b.	Determine the inter planer spacing (2 0 0), (2 2 0) and (1 1 1) planes in FCC crystal having an atomic radius of 1.246A. (CO1)	6
3-c.	What is meant by cold working and hot working of metals? Explain in detail. (CO2)	6
3-d.	What are the characteristics of pearlite and austenite. (CO2)	6
3.e.	Explain the Flame Hardening heat treatment. (CO3)	6
3.f.	What is the difference between the composite materials and the alloy? (CO4)	6
3.g.	What are the informations revealed by the microstructure examination? (CO5)	6
<b>SECTIO</b>	$\overline{\mathbf{N-C}}$	50
4. Answe	er any <u>one</u> of the following:-	
4-a.	Derive an expression for Atomic Packing Factor in case of FCC unit cell. (CO1)	10
4-b.	Define the following mechanical properties tensile strength, modulus of elasticity, toughness, and hardness. (CO1)	10
5. Answe	er any <u>one</u> of the following:-	
5-a.	Draw the iron-carbon equilibrium diagram and explain it. (CO2)	10
5-b.	What is lever phase rule to determine the mass fraction of the phases present in a binary phase diagram? Also, derive an equation. (CO2)	10
6. Answe	er any <u>one</u> of the following:-	
6-a.	Explain the objective and the procedure of Annealing heat treatment of metals with a neat sketch. (CO3)	10
6-b.	Discuss the need of quenching process. Write the importance of correct quenching media and their types. (CO3)	10
7. Answe	er any <u>one</u> of the following:-	
7-a.	What is piezo electric materials; explain various piezo electric materials and their applications. (CO4)	10
7-b.	What is meant by reinforcement materials? Write about their classification and applications. (CO4)	10
8. Answe	er any <u>one</u> of the following:-	
8-a.	What is meant by Materials characterization? What is X-Ray diffraction? What is the need of X-ray diffraction? Explain in brief. (CO5)	10
8-b.	Explain the transmission electron microscopy with a neat sketch in details. (CO5)	10